Strictly Confidential: (For Internal and Restricted use only) Secondary School Examination-2020 Marking Scheme – SCIENCE (SUBJECT CODE: 086) (PAPER CODE : 31/4/2)

General Instructions: -

- 1. You are aware that evaluation is the most important process in the actual and correct assessment of the candidates. A small mistake in evaluation may lead to serious problems which may affect the future of the candidates, education system and teaching profession. To avoid mistakes, it is requested that before starting evaluation, you must read and understand the spot evaluation guidelines carefully.**Evaluation is a 10-12 days mission for all of us. Hence, it is necessary that you put in your best efforts in this process.**
- 2. Evaluation is to be done as per instructions provided in the Marking Scheme. It should not be done according to one's own interpretation or any other consideration. Marking Scheme should be strictly adhered to and religiously followed. However, while evaluating, answers which are based on latest information or knowledge and/or are innovative, they may be assessed for their correctness otherwise and marks be awarded to them. In class-X, while evaluating two competency based questions, please try to understand given answer and even if reply is not from marking scheme but correct competency is enumerated by the candidate, marks should be awarded.
- 3. The Head-Examiner must go through the first five answer books evaluated by each evaluator on the first day, to ensure that evaluation has been carried out as per the instructions given in the Marking Scheme. The remaining answer books meant for evaluation shall be given only after ensuring that there is no significant variation in the marking of individual evaluators.
- 4. Evaluators will mark($\sqrt{}$) wherever answer is correct. For wrong answer 'X"be marked. Evaluators will not put right kind of mark while evaluating which gives an impression that answer is correct and no marks are awarded. This is most common mistake which evaluators are committing.
- 5. If a question has parts, please award marks on the right-hand side for each part. Marks awarded for different parts of the question should then be totaled up and written in the left-hand margin and encircled. This may be followed strictly.
- 6. If a question does not have any parts, marks must be awarded in the left-hand margin and encircled. This may also be followed strictly.
- 7. If a student has attempted an extra question, answer of the question deserving more marks should be retained and the other answer scored out.
- 8. No marks to be deducted for the cumulative effect of an error. It should be penalized only once.
- 9. A full scale of marks 0-80 has to be used. Please do not hesitate to award full marks if the answer deserves it.
- 10. Every examiner has to necessarily do evaluation work for full working hours i.e. 8 hours every day and evaluate 20 answer books per day in main subjects and 25 answer books per day in other subjects (Details are given in Spot Guidelines).
- 11. Ensure that you do not make the following common types of errors committed by the Examiner in the past:-
 - Leaving answer or part thereof unassessed in an answer book.
 - Giving more marks for an answer than assigned to it.
 - Wrong totaling of marks awarded on a reply.

- Wrong transfer of marks from the inside pages of the answer book to the title page.
- Wrong question wise totaling on the title page.
- Wrong totaling of marks of the two columns on the title page.
- Wrong grand total.
- Marks in words and figures not tallying.
- Wrong transfer of marks from the answer book to online award list.
- Answers marked as correct, but marks not awarded. (Ensure that the right tick mark is correctly and clearly indicated. It should merely be a line. Same is with the X for incorrect answer.)
- Half or a part of answer marked correct and the rest as wrong, but no marks awarded.
- 12. While evaluating the answer books if the answer is found to be totally incorrect, it should be marked as cross (X) and awarded zero (0)Marks.
- 13. Any unassessed portion, non-carrying over of marks to the title page, or totaling error detected by the candidate shall damage the prestige of all the personnel engaged in the evaluation work as also of the Board. Hence, in order to uphold the prestige of all concerned, it is again reiterated that the instructions be followed meticulously and judiciously.
- 14. The Examiners should acquaint themselves with the guidelines given in the Guidelines for spot Evaluation before starting the actual evaluation.
- 15. Every Examiner shall also ensure that all the answers are evaluated, marks carried over to the title page, correctly totaled and written in figures and words.
- 16. The Board permits candidates to obtain photocopy of the Answer Book on request in an RTI application and also separately as a part of the re-evaluation process on payment of the processing charges.

	MARKING SCHEME – CLASS X SCIENCE (2019-20)				
S.NO	QUESTION PAPER CODE : SET 31/4/2 VALUE POINTS/EXPECTED ANSWER	MARKS	TOTAL		
			MARKS		
1	SECTION A	1	1		
1.	Oils containing unsaturated fatty acids/name of any edible oil.	1	1		
2	The electric current generated / induced in a conductor by changing magnetic field around it.	1	1		
3.	(a) The properties of elements are the periodic functions of their atomic masses.(b) To fill with undiscovered elements.	1			
	 (c) (ii)/RH₄, RO₂ (d) (i)/Atoms of an element with similar chemical properties but 	1			
	different atomic masses.	1	4		
4.	(a) Use of separate bins for plastic and paper ; separation of biodegradable and non biodegradable waste or any other.(b)	1/2+ 1/2			
	 Packaging of articles like water , food, milk, biscuits etc. Disposable utility items- bowls, tumblers, plates, leaves etc. (c) By providing cloth /jute /earthern pots and utensils/ paper or any 	1/2 + 1/2			
	other material for the similar purposes. (d)	1			
	 Yes The action of microbes is tested in the laboratory creating the same conditions as in the landfill. 	1/2 1/2	4		
5.	(B) /fusion of nuclei of male and female gamete.	1	1		
6.	(B) / Nephron (B) / False feet developed in some unicellular organisms.	1			
7.	(C) / Valves ensure that the blood does not flow backwards.	1	1		
8.	(B) / The nucleus of Uranium is bombarded with high energy neutrons. OR (A) / Biomass	1	1		
9.	(A) /1 Ω OR	1	1		
	(B) / half	1			
10.	(C) /direction of the induced current.	1	1		
11.	(B) 10 %	1	1		
12.	Note: Treat all answers as correct. Give full credit even if not attempted.	1	1		
13.	(c) $/$ (A) is true, but (R) is false.	1	1		
14.	(a) / Both (A) and (R) are true and (R) is the correct explanation of assertion (A).	1	1		
15	SECTION B	1			
15.	(a) Double displacement reaction.(b) Due to the formation of colourless sulphuric acid in the solution.	1 1⁄2			

		~ ~			
	(c) Copper Sulphide / C			1/2	
	$(d)CuSO_4 + H_2S \rightarrow Cu$			1	3
		OR			
	(a) Electrolysis of wate			1	
	(b) Carbon electrodes /	Graphite rod		1⁄2	
	(c) $2H_2O \xrightarrow{\text{Electric}} 2H_2 \cdot$	⊥ O.		1	
	$\begin{array}{c} (C) & 2\Pi_2 \\ Current \end{array}$	F 0 ₂			
	(d) Pure water is a non	conductor of electricit	v /electric current is		
		acidic solution by ion		1/2	
16.	(a) Behind the mirror	acture solution by for	5.	1/2	
10.					
	(b) Magnified			1/2	
	(c) Virtual and erect			1/2	
	Labelled new diagnom				
	Labelled ray diagram				
			Ä		
		J.		11/	2
		Art		11/2	3
	C F		e e		
		10 cm			
		- Isom p			
17.	- Evolution Ca	adual ahanga in lining	anaaniama with times air as		
1/.	• Evolution – Gr	adual change in living	organisms with time since		
	the beginning of	of life resulting in the f	ormation of a new species/		
	Evolution is sin	mply the generation of	f diversity and the shaping		
		environmental selectio		1	
			rogress because more and		
			merged and evolved over		
	-	s does not mean	-		
			oler organisms but some		
		-	like hot springs, deep sea,	2	3
		nd the ice in Antartica			
	thermal vents a	OR			
	Examples of feathers				
	-		ulation in cold weather but		
		t become useful for flig		1/2	
		s have feathers but the		1	
				1	
		ted the feathers for fli		1/2	
10	- This shows tha	t birds are closely relat	eu to reputes.		
18.	Clanda	Location	Function		
	Glands				
	a) Pituitary gland	Brain	Stimulates growth in		
			all organs.		
	b) Thyroid	Neck/ Larynx	Regulates		
			metabolism of fats,		
			proteins &		
			carbohydrates.		
	c) Pancreas	Below Stomach	Regulates blood		
			sugar	½×6	3
		·			
31///	·	Page 1 o	<u> </u>		

19.						
	Galvan	isation		Alloying		
	1. Coating a laye	er of zinc metal	1. Mixing	of a metal with metal or		
	on the metal.		non-me			
	2. Not a homoge	neous mixture.	2. Homoge	eneous mixture.		
	3. No change in	physical	3.Change i	in physical properties of		
	properties of me	tals takes place.	metals t	akes place.		
	4. The process is	an outcome of	4. Reactivi	ty of metals do not play		
	the reactivity of		any role			
	5. Prevents rustin	ng only.		loys may prevent		
			-	also used for other		
			advantages		1 × 3	3
				(Any Three)	1 ^ 5	5
			OR			
		Cold W	ater	Hot Water		
		• Reacts vi		• React more violently		
	C - L'arre	• Heat is e	volved.	• More heat is		
	Sodium			evolved.		
		Ponote la	ess violently	(Any one point)Reacts violently with		
	Calcium	as compa	•	hot water and sticks		
		sodium.		to surface of metal		
				and floats on surface		
				of water.		
	Magnesium		react with	• React with hot water		
		cold wate	er	and floats on surface of water.		
				of water.	1 × 3	
				II		
20.	Carbon ato	om, the first mem	ber of grou	p 14 has the smallest size		
				orce of attraction.		
	-			es four more electrons to		
		le configuration.	1			
		-	us of carbon	is able to hold the		
		rs of electrons str				
				f the same group are		
		e to bigger size o		S.	1. 0	2
				(Any Three)	1×3	3

21.	The movement of the growth of the roots downwards and the shoots upwards under the stimuli of gravity is called geotropism./ The movement of the part of the plant towards or away from the stimulus gravity.	1	
	Diagram Labelling	1 1⁄2+ 1⁄2	3
22.	(a) Medium P is optically denser than Q because in it the ray is bending more towards the normal as $r = r / r$	1	
	more towards the normal as $\angle r_P < \angle r_Q$ (b) 1 dioptre is the power of a lens whose focal length is 1 metre.	1 1	
	(c) f (m) = $\frac{1}{P(D)}$	1/2	
	$=\frac{1}{+0.5\mathrm{D}}$	1⁄2	
	= +2m		3
23.	By placing second (identical) prism in an inverted position with respect to the first prism.	1	
	White light R R R R White light P_1 A		
	Diagram Labelling	1 ¹ /2 ¹ /2	3
24.	(a) Presbyopia	1⁄2	
	(b) Gradual weakening of the ciliary muscles of the eye / diminishing flexibility of the eye lens.	1	
	(c) Bifocal lens	1/2	
	Lens Lens	1	3
25.	• Homogeneous mixture of two or more metals, or a metal and a	1	
	non-metal.	-	

	elements in it in definite proportions. It is then cooled at room		
	temperature.	2	
	• Stainless steel- Iron, Nickel, Chromium	1	
	• Hard, does not rust.	1	5
26.	• Male germ-cell Pollen tube Female germ-cell		
	Labelling Diagram	1 ½ 1 ½	
	 Process of fertilization : Pollen tube is formed from the pollen grain . Fusion of male germ cell with female germ cell to form zygote. 	1	
	 Ovary- Changes into fruit Ovule- Changes into Seed 	1/2 1/2	5
	OR		
	 (a) The period during adolescence is called puberty/age of males and females at which reproductive organs become functional. (b) (i) Testes – Production of sperms /Secretion of male sex hormone testosterone. (ii) Seminal vesicle – secretes a fluid which makes the transport of sperms easier/ the fluid secreted gives nutrition to sperms. (iii) Vas deferens- carries the sperms to the seminal vesicle. 	1	
	 (iii) Vas derefens- carries the sperms to the seminar vestere. (iv) Urethra- forms a common passage for both the sperms and urine. (Any one) (c) because the sperm formation requires a lower temperature than the normal body temperature. 	½ × 4	
	(d) With the help of a long tail.	1	
27.	$2 \vee x 3 = 6 \vee key$ $+ HHH= ()$ $2 \circ 4 + 0 + 0 + 0 + 0 + 0 + 0 + 0 + 0 + 0 +$	2	

	(a) Three resistors are connected in parallel hence voltage across each is same i.e. 6V.		
	$I_1 = \frac{V}{R_1} = \frac{6}{10} = 0.6 \text{ A}$	1/2	
	$I_2 = \frac{V}{R_2} = \frac{6}{20} = 0.3 \text{ A}$	1⁄2	
	$I_3 = \frac{V}{R_3} = \frac{6}{30} = 0.2 \text{ A}$	1⁄2	
	b) I= I ₁ + I ₂ + I ₃ = 1.1 A	1⁄2	
	c) $R_{eff} = \frac{V}{I}$	1⁄2	
	$=\frac{6}{1.1}=5.4\Omega$	1/2	
	OR		
	$R_1 = R_2 = 15\Omega \qquad V = 6V$		
	i) In series : $R_s = R_1 + R_2 = 15\Omega + 15\Omega = 30\Omega$	1/2	
	$I = \frac{V}{R_3} = \frac{6V}{30\Omega} = 0.2 A$	1/2	
	$\therefore P_1 = VI = 6V \times 0.2 A = 1.2 W$	1	
	ii) In parallel		
	$R_{p} = \frac{R_{1} \times R_{2}}{R_{1} + R_{2}} = \frac{15 \times 15}{15 + 15} = \frac{225}{30} = 7.5\Omega$	1⁄2	
	$I = \frac{V}{R_p} = \frac{6V}{7.5\Omega} = 0.8 \text{ A}$	1/2	
	$P_2 = VI = 6V \times 0.8 A = 4.8W$	1	
	Ratio of power = $\frac{P_1}{P_2} = \frac{1.2W}{4.8W} = \frac{1}{4}$	1⁄2	
	$\therefore P_1: P_2 = 1:4$	1⁄2	5
28.	Olfactory indicator (a) Colourless and Odourless gas is evolved with bubbles .	1 1	
	Zinc + Acid → Zinc Salt + H_2 ↑		

	(or by using any example of acid e.g. HCl/H_2SO_4)	1	
	(b) Brisk effervescence/ colourless and odourless gas is evolved.	1	
	Sodium carbonate + Acid - Sodium salt of Acid + Water + Carbon		
	dioxide ↑	1	
	(or by using any example of acid like HCl/H_2SO_4)		5
	OR		
	• Water of crystallization is the fixed number of water molecules		
	present in one formula unit of a salt.	1	
	-	1 1⁄2	
	• Examples $CuSO_4$.5H ₂ O	⁷² ¹ / ₂	
	$Na_2CO_3.10 H_2O$ (or Any other)	72	
	• Heat a few crystals of hydrated copper sulphate(bluecolour)		
	in a dry boiling tube.	1/2	
	Water droplets are seen in the boiling tube.	1/2	
	Colour : The colour of copper sulphate changes to white .	1	
	State : The blue crystal changes to white powder.	1	
29.	(a) For providing energy for various metabolic processes / Formation of		
	new cells / Repair of damaged or worn out cells & tissues / Developing		
		1+1	
	resistance against diseases. (Any Two)	1+1	
	(b) Peristaltic movement / Peristalsis/ Rhythmic contraction and	1	
	relaxation of the muscles in the lining of alimentary canal.		
	(c) Herbivores eat plant matter which is rich in cellulose and takes		
	longer time to digest and hence longer small intestine.	1	
	(d) The inner lining of the stomach will not be protected from the action	1	
	of the acid /HCl	1	5
30.		1/-	5
50.	(a) Mode : Parallel arrangement	1/2	
	Reasons :		
	• Voltage across each appliance is same.		
	• Selective operation of devics is possible /separate switch can be		
	provided in the circuit for each appliance.		
	• If one device is defective or non-operational, working of others		
	is not affected.		
	(Any Two)	1×2	
	(b) 5A	1/2	
	15 A	1/2	
	Due to difference in power rating of appliances used in household.	1/2	
	(a) When live wine and neutral wine some in direct contest		
	(c) When live wire and neutral wire come in direct contact/	1 /	
	overloading.	1/2	
	Prevention :		
	By the use of fuse/ using good quality cables.	1/2	5
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